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**Tesis: "SYNTHESIS AND CHARACTERIZATION OF OXIDES AND FLUORIDES DOPED NANOCRYSTALS WITH UV-VIS-NIR EMISSION"**

**Resumen:**

In this thesis is presented a study of structural, morphological and optical properties of lanthanide-doped nanocrystalline oxides and fluorides. Ytterbium ( $\text{Yb}^{3+}$ ), Erbium ( $\text{Er}^{3+}$ ), Gadolinium ( $\text{Gd}^{3+}$ ), Bismuth ( $\text{Bi}^{3+}$ ), Boron ( $\text{B}^{3+}$ ), Europium ( $\text{Eu}^{3+}$ ), Holmium ( $\text{Ho}^{3+}$ ), and praseodymium ( $\text{Pr}^{3+}$ ) doped and co-doped  $\text{ZrO}_2$ ,  $\text{Y}_2\text{O}_3$ ,  $\text{Y}_2\text{O}_2\text{S}$ , and  $\text{NaYF}_4$  were studied under UV and IR excitation. Most lanthanide-doped oxides were prepared by precipitation method combined with a hydrothermal process, and a few by combustion method. The crystal structure was investigated by X-Ray diffraction, and it was obtained tetragonal phase, cubic phase, and hexagonal phase for the lanthanide-doped and codoped  $\text{ZrO}_2$ ,  $\text{Y}_2\text{O}_3$  and  $\text{Y}_2\text{O}_2\text{S}$ , respectively; two mixed phases, cubic and hexagonal were presented in lanthanides  $\text{NaYF}_4$ .  $\text{ZrO}_2:\text{Yb}^{3+}$ ,  $\text{Er}^{3+}$ ,  $\text{Gd}^{3+}$  nanocubes were synthesized with  $\text{Na}_2\text{S}$ , where the enhancement of the UC signal with the use of  $\text{Na}_2\text{S}$  was the result of the substitution of  $\text{OH}^-$  by  $\text{SO}_4^{2-}$  ions on the surface of nanoparticles. While the strong enhancement observed with the introduction of  $\text{Gd}^{3+}$  was the result of a forward and backward energy transfer process ( $\text{Er}^{3+} \rightarrow \text{Gd}^{3+} \rightarrow \text{Er}^{3+}$ ). Tridoped  $\text{ZrO}_2:\text{Yb}^{3+}-\text{Er}^{3+}-\text{Bi}^{3+}$  nanophosphors were synthesized and demonstrated the possibility to switch from green-yellow to red emission by switching the excitation from 350nm to 970nm. It was obtained  $\text{Y}_2\text{O}_3:\text{Yb}^{3+}$ ,  $\text{Er}^{3+}$ ,  $\text{M}^{3+}$  ( $\text{M}^{3+}$ :  $\text{Bi}^{3+}$ ,  $\text{Gd}^{3+}$ ,  $\text{Eu}^{3+}$ ,  $\text{Ho}^{3+}$ ,  $\text{Pr}^{3+}$ ) nanoparticles with cubic crystalline phase, under precipitation method combined with a hydrothermal process, and combustion synthesis. Besides enhancing considerably the intensity of luminescence,  $\text{Bi}^{3+}$  and  $\text{Gd}^{3+}$  present own emissions. On the other hand, combining  $\text{Eu}^{3+}$  and  $\text{Ho}^{3+}$ , or  $\text{Yb}^{3+}$  and  $\text{Pr}^{3+}$ , it is possible to obtain visible emission in different wavelengths changing the excitation source. Through a modified homogeneous precipitation method combining low-temperature sulfurization process at 625 °C, it was possible to get  $\text{Y}_2\text{O}_2\text{S}:\text{Yb}^{3+}$ ,  $\text{Er}^{3+}$ ,  $\text{M}^{3+}$  with pure hexagonal phase, which favorably enhanced the luminescence intensity of nanoparticles by a factor of 33. Most SEM images show uniformly bonded plates forming like a flower with tendency to spherical shape.  $\text{NaYF}_4:\text{Yb}^{3+}$ ,  $\text{Er}^{3+}$  nanoparticles were synthesized by co-precipitation method. DRX patterns showed mixed phases corresponding to cubic and hexagonal phase at 600 °C. SEM images showed dispersible nanoparticles with smooth and well-formed crystalline forms, most of them had polyhedral shaped blocks with size around 90 nm. It was observed visible (VIS) and near infrared emission (NIR). The color emission could be observed under excitation of a portable laser pointer. Most of synthesized nanoparticles in this thesis could be excited either with IR or UV source in order to get visible emission with a specific primary color (red and green). Interestingly, some nanoparticles show UV and NIR emission. All results demonstrated in this thesis place these nano-oxides and nano-fluorides as excellent candidates in biolabeling as markers for multiplexed detection due to their tuning color properties, in security as anti-counterfeiting effect on inks due to their widely emission range (UV, VIS and NIR), and for this last emission in optical communications.